Direct Determination of Total Mercury in Mineral Coal Samples

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Abstract
Mercury is recognized as a potent neurotoxic substance and it is included in the list of Persistent Toxic Substances\(^1\). It is estimated that anthropic emissions of atmospheric Hg are around 2,000 t year\(^{-1}\), which 25\% is originated from coal burning\(^2\). In Brazil, MMA calculated emissions of 1.0 t of Hg year\(^{-1}\) due to coal burning in thermoelectric plants\(^3\). However, data about Hg concentrations in Brazilian coal are scarce. This work aimed at the determination of Hg, based on thermal decomposition, amalgamation and detection by AAS. This analytical method of solids is efficient, free of sample preparation and avoids loss and contamination. The influence of the analyzed mass sample and the matrix effect has been evaluated. Method validation has been performed by the mean of analytical curve linearity, accuracy, precision, limit of detection and limit of quantification. The validated method was applied in coal samples from different regions of Brazil and of the world and the results showed mercury concentration values close to the values found in the literature.

Key words: Mercury, Mineral Coal, Direct Analyzer.

Introduction
The aim of this work was to optimize and validate a method for total Hg determination in coal using a Direct Mercury Analyzer (DMA-80, TRICELL, Milestone, Italy).

Results and Discussion
The analytical curve presented good linearity between 0.1 and 20 ng, with a correlation coefficient \((R^2)\) > 0.9986 in Cell 0 and Cell 1, and residual sum of squares of \(3.42 \times 10^4\) and \(2.03 \times 10^3\) respectively for both cells. It was performed a Student’s t-test and it was not verified significant difference between the angular coefficients of aqueous solutions curve and curve built with standard addition in coal sample \((p > 0.05)\). The method was considered exact, with recovery of 107\% of the certificate value (Coal SRM 1632d). An addition and recovery study confirmed the method accuracy, being verified recoveries of 95\% and 101\%, for Hg additions of 1 ng and 5 ng, respectively. Precision is related with sample granulometry. The decrease of the granulometry improves the precision, due to a better homogenization of the sample. It was obtained a Relative Standard Deviation > 39\% and < 6\%, for granulometry > 500 and < 200 \(\mu\)m, respectively. In mass interval between 5 and 30 mg, it was not found significant variation (Student’s t-test, \(p > 0.05)\) of the Hg concentration. The limit of detection and quantification were 0.017 ng and 0.059 ng, respectively. The validated method was applied in mineral coal sample from Brazil, Canada, Colombia, Mozambique and USA. The Hg concentration for Brazilian samples ranged of 0.132 to 0.293 mg kg\(^{-1}\) and the Hg concentration of all the regions was of 0.056 to 0.293 mg kg\(^{-1}\).

Conclusions
The overall Hg concentrations obtained in this work are in accordance with previous data (0.02 to 1 mg kg\(^{-1}\))\(^4\) and Hg concentrations found for Brazilian samples are in the range of 0.05 to 0.8 reported by Pires et al.\(^5\) for Brazilian coal. The analysis of more coal samples will allow the formation of a more consistent database. This data is necessary to calculate Hg emission factors of thermoelectric plants.

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\(^1\) UNEP (United Nation Environmental Program), Mercury, time to act, UNEP, Geneva, Switzerland, 2013a, 44 p.
\(^2\) UNEP, Global Mercury Assessment: sources, emissions, releases and environmental transport, 2013b.
\(^3\) MMA (Ministério do Meio Ambiente), Diagnóstico Preliminar sobre o Mercúrio no Brasil, Ministério do Meio Ambiente, Brasília, Brasil, 2013, 106p.